

PERIODIC TABLE OF THE ELEMENTS

<http://www.ktf-split.hr/periodni/en/>

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IIA

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LITHIUM

4

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Be

BERYLLIUM

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III A

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B

BORON

16

III A

6

12.011

C

CARBON

7

14.007

N

NITROGEN

8

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O

OXYGEN

9

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F

FLUORINE

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22.990

Na

SODIUM

12

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Mg

MAGNESIUM

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26.982

Al

ALUMINIUM

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28.086

Si

SILICON

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P

PHOSPHORUS

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S

SULPHUR

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Cl

CHLORINE

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K

POTASSIUM

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Ca

CALCIUM

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Sc

SCANDIUM

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Ti

TITANIUM

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V

VANADIUM

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Cr

CHROMIUM

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Mn

MANGANESE

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Fe

IRON

27

58.933

Co

COBALT

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COPPER

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POLONIUM

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At

ASTATINE

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FRANCIUM

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RADIUM

89-103

Ac-Lr

Actinide

104

(261)

Rf

RUTHERFORDIUM

105

(262)

Db

DUBNIUM

106

(266)

Sg

SEABORGIUM

107

(264)

Bh

BOHRIUM

108

(277)

Hs

HASSIUM

109

(268)

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MEITNERIUM

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UNUNNIUM

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UNUNOCTIUM

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LANTHANIDE

57 138.91 La LANTHANUM	58 140.12 Ce CERIUM	59 140.91 Pr PRASEODYMIUM	60 144.24 Nd NEODYMIUM	61 (145) Pm PROMETHIUM	62 150.36 Sm SAMARIUM	63 151.96 Eu EUROPIUM	64 157.25 Gd GADOLINIUM	65 158.93 Tb TERBIUM	66 162.50 Dy DYSPROSIUM	67 164.93 Ho HOLMIUM	68 167.26 Er ERBIUM	69 168.93 Tm THULIUM	70 173.0 Yb YTTERBIUM
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ACTINIDE

89 (227) Ac ACTINIUM	90 232.04 Th THORIUM	91 231.04 Pa PROTACTINIUM	92 238.03 U URANIUM	93 (237) Np NEPTUNIUM	94 (244) Pu PLUTONIUM	95 (243) Am AMERICIUM	96 (247) Cm CURIUM	97 (247) Bk BERKELIUM	98 (251) Cf CALIFORNIUM	99 (252) Es EINSTEINIUM	100 (257) Fm FERMIUM	101 (258) Md MENDELEVIUM	102 (259) No NOBELIUM
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(1) Pure Appl. Chem., 73, No. 4, 667-683 (2001)

Relative atomic mass is shown with five significant figures. For elements having no stable nuclides, the value enclosed in brackets indicates the mass number of the longest-lived isotope of the element.

However three such elements (Th, Pa, and U) do have a characteristic terrestrial isotopic composition, and for these an atomic weight is tabulated.

Editor: Aditya Vardhan (adivar@netlinx.com)

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